CZECHOSLOVAK FOOTPRINTS IN THE DEVELOPMENT OF METHODS OF THERMOMETRY, CALORIMETRY AND THERMAL ANALYSIS

A tribute to professor Vladimír Šatava, DrSc, a mastermind of theoretical basis of thermal analysis, who celebrates his 90th birthday dedicating also the 55th anniversary since he becomes the Editor of the journal Ceramics-Silikáty

*PAVEL HOLBA, JAROSLAV ŠESTÁK

New Technologies - Research Centre of the Westbohemian region, University of West Bohemia in Pilsen, Universitní 8, 301 14 Plzeň, Czech Republic

#E-mail: holbap@gmail.com

Submitted March 15, 2012; accepted May 16, 2012

Keywords: History, Thermal analysis, Calorimetry, Kinetics, DTA

A short history on the development of thermometric methods are reviewed accentuating the role of Rudolf Bárta in underpinning special thermoanalytical conferences and new journal Silikáty in fifties as well as Vladimir Šatava mentioning his duty in the creation of the Czech school on thermoanalytical kinetics. This review surveys the innovative papers dealing with thermal analysis and the related fields (e.g. calorimetry, kinetics) which have been published by noteworthy postwar Czechoslovak scholars and scientists and by their disciples in 1950-1980. Itemized 227 references with titles show rich scientific productivity revealing that many of them were ahead of time even at international connotation.

Historical roots of thermal sciences - from thermoscopy to thermal analysis

One of the first modern-times considerations of heat and cold can be found in the treatise published in 1563 by B. Telesio. At the end of the 16th century the first air thermoscope appeared (G. Galileo about 1597) and in 1626 the word "thermometer" was for the first time used to describe thermoscope equipped with scale with eight degrees (Leurechon in book "La Recreation Mathematique"). Shortly after, the world-known Czech educator J. A. Comenius inserted reflections on the role of heat and cold in nature into his work "Physicae Synopsis" (1633) and then, in 1659, published another worth noting book "Disquisitiones de Caloris Frigoris et Natura." The first quantitative thermal law expressing the dependence of temperature of a cooling body (expressed in the scale of 8 degrees) on the time was published by *I*. Newton in 1701.

Meanwhile, other scientists had invented various types of the dilatation thermometers and had proposed various temperature scales. *Rømer* (1701) had filled glass tube of thermometer with red wine and proposed a 60-degree scale. *Fahrenheit* (1724) proposed a temperature scale of 100 degrees from 0°F (at the temperature of mixture of ammonium chloride, water and ice) and

100°F at the human body temperature. *Reaumur* (1731) introduced the temperature scale with 80 degrees between 0°Re at water melting point and 80°Re at boiling point of water. A year later *Delisle* introduced an exotic scale with 240 degrees, which was later (1738) modified and adjusted to 150°D corresponding to the melting point of water and to 0°D at boiling point of water (240°D = 60°C), and this scale was being used in Russia for the whole hundred years. Only then *Celsius* (1742) came with its 100-degree scale (between the melting 100 and boiling 0 points of water, later switched to nowadays 0-100 by *Linné*).

The crucial experimental studies, thanks to which temperature became a clearly measurable physical quantity, was executed by *Regnault* in the 1840th, that is long after the *Black* (1761) distinguished between the specific heat (heat capacity) and the latent heat, *Laplace* and *Lavoisier* (1786) performed their first calorimetric measurements. In 1822 *Fourier* published his laws of heat transfer. Yet after detailed results of *Regnault's* dilatometric and heat capacity measurements (1842), together with *Carnot's* theorem (1824) and its consequent interpretation by *Clapeyron* (1834) - the basis was formed for the introducing of absolute temperature scale by *W. Thomson (Kelvin* 1848) and for the inception of thermodynamics as a new science.

The first noted use of thermometry as a method of thermal analysis took place in Uppsala in 1829 where *F. Rudberg* (1800-1839) recorded inverse cooling-rate data for various alloys. [1,2]. In 1883, *H. L. Le Chatelier* (1850-1936) adopted a somehow more fruitful approach plotting the time vs. temperature curves easily convertible to the relation of sample temperature vs. environmental temperature.

Several years later Le Chatelier (1887) had used Pt/PtRh thermocouple and the new era of thermometry as well as of calorimetry has arrived. In 1891, W. C. Roberts-Austen (1843-1902) [3] became known to construct a device to give a continuous record of the output from thermocouple and he termed it as "thermoelectric pyrometer". Later with his assistant A. Stanfield published in 1899 heating curves for gold, which almost stumbled upon the idea of DTA ("Differential Thermal Analysis"). They improved the sensitivity by maintaining the thermocouple 'cold' junction at a constant temperature and by measuring the differences between two high temperatures [4]. Among other well-known inventors was Russian N. S. Kurnakov (1860-1941) improving registration of his pyrometer by the photographic continuously recording drum [5]. The term "thermal analysis" was coined by G. H. J. Tammann (1861-1938) [6] around the year 1904 demonstrating the significance of cooling curves in phase-equilibrium studies of binary systems.

The first Czech university textbook on the physics of heat was "Thermika" by Č. Strouhal (1850-1922) published in 1908 [7] maintaining its informative value until today's. The historical development and practical use of DTA in the territory of former Czechoslovakia [1, 2, 8] was linked with the names J. Burian (1873-1942), O. Kallauner (1886-1972) and J. Matějka (1892-1960) who introduced thermal analysis as the novel technique during the period of the so called "rational analysis" of ceramic raw materials [9] in order to investigate behavior of kaolinite [10, 11] at heating.

Worth a special attention is an original development of weight measurements that is connected with the name S. Škramovský (1901-1983), who, at the Charles University, investigated thermal decomposition of complex oxalates which led him in 1932 to his own construction of an apparatus named "stathmograph" (from Greek "stathmos" = mass, weight) [12] that made it possible to measure mass changes. Independently (twenty years later), C. Duval used for his way of weight measurements the Latin-based term "thermogravimetry" that later became generally accepted in thermal analysis [13]. As the principle scheme of the stathmograph instrument is not generally known, it is perhaps worth mentioning to describe this early arrangement. Škramovský placed a weighted sample into the drying oven on a dish suspended on a long filament passing through a hole in its upper wall (forming the balance case) and hooked to an arm of an analytical balance.

A mirror attached to the beam was reflecting the image of alight slit into a slowly rotating drum lined with photosensitive paper. The unwanted vibration was reduced by an attached glass rod immersed into paraffin oil and temperature was registered automatically by means of a mercury thermometer provided by platinum contacts distributed along the whole length of capillary.

In the first years after World War II the other monographs appeared in the world literature (besides that by Duval [13]), which were devoted to microcalorimetry [14] and thermal analysis [15, 16] and an initial paper dealing with theory of DTA [17] was also published. At the end of 1950th a commercial device combining DTA and TG appeared under the name "Derivatograph" [18] for long providing a useful service to the Eastern scientists.

Much credit for the further development of modern thermal analysis was attributed with Rudolf Bárta (1897-1985) who stimulated thermal analysis activity at his fellow workers (V. Šatava, S. Procházka, J. Vašíček, M. *Čáp or I. Proks*) at the Institute of Chemical Technology Prague (abbreviated as VŠChT) [19, 20, 21, 22]. Bárta organized premature thermoanalytical meetings, the earliest was "the 1st Conference on DTA" (Prague 1956), the 2nd (Prague 1958) and the 3rd Conference on Thermography (Prague 1961) followed by the 4th Conference on DTA (Bratislava 1966). His friend R. C. Mackenzie (1920-2000) [23, 24] from Scotland was an invited guest at the 1961 meeting who was also one of the pioneers of applied DTA [23, 24]. Upon the previous communication with Russian L.G. Berg, Americans P.D. Garn and C. B. Murphy as well as Hungarian L. Erdey an idea for the creation an international society ICTA was cultivated and realized during the first international thermoanalytical conference in London 1965 [24] (where one of the authors also participated as invited speaker). It aimed to enabling easier contacts between national sciences, particularly across the separating 'iron curtain", which in that intricate time politically divided the East and West Europe.

Besides significance of the early Czech-written books on thermal science [7, 26], which appeared before and/or simultaneously with the credited international literature [8, 22] the indispensable figure in the Czechoslovak development of thermal analysis was undoubtedly Vladimír Šatava (*1922) [27-31]. He brought to the Czech scientific circles necessary theoretical basis on solid-state chemistry and physics [29, 30], pioneered methods of thermal analysis [18, 19, 27, 26], educating his students who thus followed his professional guidance and published esteemed books [27-33] completing thus the rich spectrum of Czech thermoanalytical literature [31-38], cf. Fig. 1. The individual contributions and innovative approaches have been affluent and focal which is worth of a more detailed portrayal, as exposed in the next paragraph, especially accentuating Czechoslovak source journals. Unfortunately, most of these rather crucial papers disappeared in the shadows of time due to the Czech written texts. Only consequently a supply role started playing the novel thermoanalytical journals, Thermochimica Acta, which was cofounded by one of the authors back in the year 1970 as well as Journal of Thermal Analysis instigated by R. Bárta in 1969.

Less known book by J. A. Komenský "Investigation of the nature of heat and cold." (Amsterodam 1659) in which the predicament of heat and cold is well discussed; "Thermics" by Č. Strouhal (1908) was an unique book describing the early but elementary treaties on heat, the almost unknown book on DTA (1957) was published ahead of time, basic book of solid-state chemistry and material thermal behavior was also published (1965) beforehand of international literature (unfortunately never translated). Far right is the Russian translation of Czech original book on theoretical basis of thermal analysis (1988), which became curiously a scientific bestseller as whole 2000 issues were sold in the former USSR within one week.

Methodical footpath identifiable on the territory of former Czechoslovakia

The greatest promotion of thermoanalytical methods came after fifties when the methodical bases were formed [39] and new techniques specified. In this period various Czechoslovak scientists played an important role as it is documented in the achievement book [40] and citation records [41]. Below listed papers relating the field of TA promoted in the former Czechoslovakia are assorted into several (but not very strict) categories. The referenced papers are supplied by original titles (given in English) and they are chronologically ordered within individual categories. All references were checked and corrected according to database WOS (Web of Science).

The first category "*TA generally*" consists of articles [42-57] deals mostly with thermal analysis in a general way including papers published mainly in Czech journals Silikáty and Chemické listy, and in Slovak

journal Chemické Zvesti.

The second category "Special methods of TA" is devoted to original principles and unique techniques developed and put into operation by Czechoslovak scientists. The articles described e.g. dielectric TA [58], thermogravimetry [59, 60, 63], accelerated TA [62], permeability TA [65], photometric TA [66], periodic TA [68] (becoming a forerunner of today's temperature modulated methods of TA), differential hydrothermal analysis [70] and [71, 74], quick TA [78], thermoelectrometry [79, 80], decrepitating TA [82] and thermomagnetometry [83]. A distinctive consideration should be allocated to the characterization of radioactive measurement called emanation thermal analysis (ETA) which is connected mainly with authors V. Jesenák [64], V. Balek [67, 75, 76] and J. Tölgyessy [81].

An explicit part of the papers was devoted to the description of own constructions of apparatuses for TA measurement as a consequence of at that time existing inaccessibility of commercial TA instruments. This type of articles is included into category *Apparatuses of TA* [84-101]. Early instruments as were opportunely produced by laboratory groundwork, such as DTA belong into this category. The production way of latterly produced TG apparatuses was paved by the development of a Czech thermogravimetric instrument named "TEGRA" and constructed by *A. Blažek* [37, 87]. Early instruments were opportunely produced by laboratory groundwork, such as DTA [84, 86, 91, 97].

A rich sphere of Czechoslovak research was also formed by calorimetric contributions [103-127] registered in category *Calorimetry*. Worth accentuating is an initial classification of calorimeters according to the temperature difference between the sample-block T_B and surrounding jacket T_J as early suggested by *J. Velíšek* [117].

Consequent category *Theory of DTA/DCS* is associated with a gradual development of theoretical basis of thermal measurements mostly focused on the DTA [128-140] curiously noting early the associated effect of gradients [129, 130]. It involved problems of

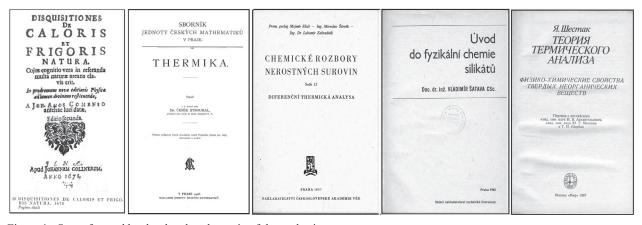


Figure 1. Some favored book related to the topic of thermal science

calibration and standardization of temperature and heat measurements by employing solid solution [135, 140], application of heat pulses [134], conductivity issues [138] as well as a detailed analysis of the complex composition of a DTA peak including the effect of heat inertia [135, 137]. One of the frequently cited and widely applied treaties was the Hrubý glassforming coefficient [133] based on the DTA determination of characteristic temperatures during glass crystallization (inquisitively becoming the best cited paper in the history of Journal Czechoslovak Physicists with 372 citations). Such achievements were only possible by the impact of prosperous Czechoslovak school on thermodynamics [27-34]. The other corresponding papers [141-162] are included into category Thermodynamics and phase equilibria

Another special attention is paid to the studies on reaction dynamics which topic is included into the category Kinetics [163-199]. Early kinetic studies were explicitly offered by studies of V. Šatava whose kinetic evaluation method [175] have been broadly exploited and quoted by international resources (several hundreds of citations) and frequently named as the "Satava kinetic method" [31, 34, 35, 174, 183]. Such a popularity of theoretical works aimed to elucidate predicaments of reaction kinetics [34, 35, 183] became the heart of the so-called Czech school of nonisothermal kinetics recently continued both in Czech [195-198] and Slovak republic [190, 191, 199]. Worth noting is the first ever published algorithm for the computer calculation of kinetic data [171] and the review paper [170] which despite the Czech language became the best cited article in the journal history. In feedback stance the papers [177, 179] undertook abundant citation responses becoming thus the best cited papers in relevant journals (562 and 132 respectively) and bringing into literature the notations named in the international literature after the authors (i.e., the Šesták-Berggren [178] as well as the Holba-Šesták [180] kinetic equations). Not less important have been the contributions by recently deceased Ivo Proks (1926-2011), who factually paved the way to the development of methods using the modulated temperature modes [68], early accounting on temperature gradients and measurement accuracy [129, 132], improved solution calorimetry [104, 113, 118] thus significantly contributing the elementary attributes of thermochemistry and thermodynamics. Worth mentioning are also his imperative studies on historical root of thermodynamics [8, 220, 222, 224, 225] as well as the work by *J. Brandštetr* [93, 100, 108, 115, 120, 125] in the field of titrimetry.

Not less important were also the related articles about mechanic properties (which was one of favored of the Šatava's research topics [208, 217]), diffusion studies [203-212] and early measurements of electrical and heat conductivity [202, 214-217] inserted into category *Mechanical and transport properties* [200-216]. The

last category of Czechoslovak papers dealing with TA is labeled as *History and nomenclature*. It contains reviews of historical aspects of thermometry and thermodynamics [221-227] and associated nomenclature issues [219, 220].

CONCLUSIONS

The Czech researches have richly contributed to thermal science and it would be a misfortune to allow their input to slip into oblivion. Clearly, one of the most important moments in the development of modern thermal analysis was the establishment of the journal Ceramics-Silikáty, launching 1956 by Rudolf Bárta. Vladimír Šatava was its chief editor within the years 1957-1967. This period was also credited with creating the foundations of thermal analysis and physical chemistry in general [29-31]. Šatava inspired his students and coworkers, cf. photo, by continuously broaden his own scientific interest in elucidating solidstate reactions which subsequently flourished into publication of various thermoanalytical books [34-39]. Equally important was the introduction of various novel thermoanalytical methods [58-83] which preceded the international know how, e.g. the emanation thermal analysis [50, 64, 67, 75, 76, 99] that has become the source of a commercially produced instrument. Specially the Czech contribution to the DTA technique [62, 71, 84-86, 91, 96, 97, 128, 131, 133, 136, 137] deserves a distinctive attention, the Czech written book published in 1957 [26] preceded international publications and was later followed by well cited Czechoslovak books [31-38]. Solid grounds of DTA were accomplished in 1976 by the consistent theory made up by Holba and Šesták and published in Ceramics-Silikáty [135, 136]. Fundamental contributions already appeared in the first issues of the journal Silikáty, Unfortunately, they did not get into a wider attention of international public due to the Czech language,. Nevertheless the Czech journal



Figure 2. Vladimir Šatava between his former students (graduating VŠChT in 1962 under his supervision) Jaroslav Šesták (left) and Pavel Holba (right) when celebrating his 89th birthday.

Ceramics-Silikáty as well as its sister's Chemické Listy and Czechoslovak Journal of Physics has remained important domestic as well as international sources and platform of many original ideas and we should be thankful to the effort of their originators as well as their current editors.

References

- Šesták J., Mareš J. J.: From caloric to statmograph and polarography; J. Thermal Anal. Calor. 88, 763 (2007).
- Šesták J., Mareš J. J., Hubík P.: Historical roots and development of thermal analysis and calorimetry; chapter 21 in the book Glassy, amorphous and nano-crystalline materials, Šesták J., Mareš J. J., Hubík P. (edts), Springer, Berlin 2011, pp. 347-370 (ISBN 978-90-481-2881-5).
- 3. Roberts-Austen W. C.: Fifth report to the alloys research; Nature 59, 566-567 (1899); and Proc. Inst. Mech. Eng. 35 (1899).
- Stanfield A.: On some improvements in the Roberts-Austen recording pyrometer, with notes on thermo-electric pyrometry; Phil. Mag. 46, 59 (1898).
- 5. Kurnakov N. S.: Eine neue Form des Registrierpyrometers; Z. anorg. Chemie 42, 184 (1904).
- 6. Tammann G.: Über die Anwendung der Thermische Analysen in abnormen Fällen; Z. anorg. Chem. 45, 24 (1905) and: Über die Anwendung der Thermische Analysen III; Z. anorg. Chem. 45, 289 (1905).
- 7. Strouhal Č.: Thermika; JČMF, Praha 1908.
- 8. Šesták J., Proks I., Šatava V., Habersberger K., Brandštetr J., Koráb O., Pekárek V., Rosický J., Vaniš M., Velíšek J.: *History of thermal analysis on the territory of former Czechoslovakia*; Thermochim. Acta *100*, 255 (1986).
- 9. Kallauner O., Matějka J.: Beitrag zu der rationellen Analyse; Sprechsaal 47, 423 (1914).
- 10. Matějka J.: Chemical changes of kaolinite on firing; Chemické listy 13, 164 (1919); and 182 (1919)
- 11. Matějka J.: Thermal analysis as a tool for determination of kaolinite in soils; Chem. Listy 16, 8 (1922).
- 12. Škramovský S.: Apparatus for automatic registration of dehydration at rising temperature; Chemické listy 26, 521 (1932).
- 13. Duval C.: *Inorganic Thermogravimetric Aanalysis*; Elsevier, Amsterdam 1953.
- Swietoslawski W.: Microcalorimetry, Reinhold, New York 1946.
- 15. Vold M. J.: Differential thermal analysis, Anal. Chem. 21, 683 (1949).
- 16. Berg L. G.: Скоростной количественный фазовый анализ; (Rapid Quantitative Phase Analysis), Akad. Nauk, Moscow 1952.
- Boersma S. L.: A theory of DTA and new methods of measurement and interpretation; J. Amer. Cer. Soc. 38, 281 (1955).
- 18. Paulik F., Paulik J., Erdey, L.: *Der Derivatograph*; Z. anal. Chem. *160*, 241 (1958); and: *Derivatographie*; Bergakademie 12, 413 (1960).
- 19. Bárta R., Šatava V.: *DTA as a quick control and investigative method in chemical industry;* Chem. prům. 3,113 (1953).
- 20. Šatava V.: Significance of DTA in the industry of cements;

- Stavivo 31, 15 (1953).
- 21. Bárta R.: *International directions for thermal analysis*; Chem. Listy *62*, 454 (1968).
- 22. Šatava V.: *Rudolf Bárta-Obituari*; Silikáty 29, 289 (1985).
- 23. Mackenzie R. C. (ed.): *The differential thermal investigation of clays;* Mineral. Society, London 1957.
- Lombardi G., Šesták J.: Ten years since Robert C. Mackenzie's death: a tribute to the ICTA founder; J. Thermal Anal. Calor. 105, 783 (2011).
- 25. Currell B. R.: Thermal Analysis; Science 13, 765 (1965).
- Eliáš M., Šťovík M., Zahradník L.: Diferenční termická analýza; (Differential Thermal Analysis), AVČR, Praha 1957.
- 27. Šatava V.: Documentation on thermal analysis; Silikáty 1, 240 (1957).
- 28. Šatava V.: Utilization of thermographic methods for studying reaction kinetics; Silikáty 5, 68 (1961).
- Šatava V.: Úvod do fyzikální chemie silikátů; (Introduction to Physical Chemistry of Silicates), SNTL, Praha 1965.
- 30. Šatava V.: Fyzikální chemie heterogenních soustav základy klasické a statistické termodynamiky; (Physical chemistry of heterogeneous systems foundation of classical and statistical thermodynamics), SNTL, Praha 1977 and: Fyzika pevných látek; (Solid state physics), SNTL 1979 and: Fyzikální chemie silikátů na bázi racionální termodynamiky; (Physical chemistry of silicates based on rational thermodynamics), SNTL 1981 and: Fyzikální chemie silikátů kinetika; (Physical chemistry of silicates kinetics), SNTL 1982.
- 31. Šesták J., Šatava V., Wendlandt W. W.: *The Study of Heterogeneous Processes by Thermal Analysis*; Monography as a special issue of Thermochimica Acta 7, 333 (1973).
- 32. Šesták J.: Měření termofyzikálních vlastností pevných látek: teoretická termická analýza; Akademia, Praha 1982 and English translation: Thermophysical Properties of Solid: theoretical thermal analysis; Elsevier Amsterdam 1984 and: Russian translation Mir, Moscow 1988.
- 33. Holba P.: *Thermodynamics and ceramic systems*; chapter 1 in book *Structure and Properties of Ceramic Materials*; (A. Koller, ed.). Elsevier, Amsterdam 1994, pp. 17-113.
- 34. Šesták J.: Science of Heat and Thermophysical Studies: a Generalized Approach to Thermal Analysis; Elsevier, Amsterdam 2005.
- Šesták J., Šimon P. eds.: Thermal Analysis of Micronano- and non-crystalline Materials: transformation, crystallization, kinetics and thermodynamics; Springer, Berlin 2012 (ISBN 978-90-481-3149-5).
- Balek V., Tölgyessy T.: Emanation Thermal Analysis and other Radiometric Emanation Methods; Elsevier, Amsterdam 1984 and: Russian translation Mir, Moscow 1987.
- 37. Blažek A.: *Termická analýza*; SNTL, Praha 1972 and: English translation: *Thermal Analysis*; Van Nostrand-Reinhold, London, 1973.
- Kubičár L.: Pulse Methods of Measuring Basic Thermophysical Parameters; Elsevier, Amsterdam 1990.
- Liptay G., (ed). Atlas of Thermoanalytical Curves: (TG, DTG, DTA curves measured simultaneously); London, New York: Heyden and Son 1971.
- Liptay G., Simon J. (eds): Who is who in thermal analysis and calorimetry; Akademiai Kiado, Budapest 2004.

41. Šesták J.: *Citation records and some forgotten anniversaries in thermal analysis;* J. Thermal Anal. Calor. in print 2012 (DOI:10.1007/s10973-011-1625-3).

TA generally

- 42. Šatava V.: Temperature regulators for thermography; Silikáty 1, 204 (1957).
- 43. Šatava V.: Thermische Zersetzung von Ammoniummetavanadat; Coll. Czech. Chem. Comm. 24, 2171 (1959).
- 44. Blažek A.: Beitrag zur experimentellen Methodik der thermische Analyse; Bergakademie 12, 191 (1960).
- 45. Číčel B.: Thermal methods of analysis; Chem. Zvesti 20, 154 (1966).
- 46. Biroš J.: Techniques and methods of polymer evaluation by TA; Chem. Listy 61, 1243 (1967).
- 47. Habersberger K.: Application of TA to investigation of catalysts; J. Thermal Anal. 12, 55 (1977).
- 48. Pospíšil Z.: Complex thermal analysis and its utilization in the field of ceramics; Silikáty 25, 263 (1981).
- Fellner P., Votava I.: Numerical treatment of cooling/ heating curves at thermal analysis; Chem. Zvesti 35, 31 (1981).
- 50. Balek V, Beckmann I.: Use of labeled atoms in thermal analysis; Chem. Listy 79, 19 (1985).
- 51. Mentlík V.: New application of DTA in heavy-current electrotechnology; Thermochim Acta 93, 353 (1985).
- 52. Holba P.: Processing of TA curves and the exact use of thermal analysis. Thermochim Acta 110, 81 (1987).
- 53. Habersberger K., Balek V.: *Present state of commercially available thermoanalytical equipment;* Thermochim Acta 110, 47 (1987).
- Kubičár L.: Trends in the methods of measurement of thermophysical properties in the solid state; Thermochim Acta 110, 209 (1987).
- 55. Šatava V., Lach V.: Thermal analysis and chemistry of inorganic binders; Thermochimica Acta 110, 467 (1987).
- Šesták J.: Non-traditional and traditional methods of thermal analysis in solid-state chemistry and physics; Thermochim Acta 148, 79 (1989).
- 57. Balek V., Karabascheva N. A., Györyová K.: *Literature* survey on thermal analysis reference materials; J. Thermal Anal. 40, 1459 (1993).

Special methods of TA

- 58. Bergstein A.: Changes during ignition of equimolecular mixtures of barium carbonate and titanium dioxide, followed by measurement of dielectric properties; Collect. Czech. Chem. Com. 20, 1041 (1955).
- Šatava V.: Simple registration thermobalance; Silikáty 1, 188 (1957).
- 60. Blažek A.: Electronic thermobalance with simultaneous recording of DTA curves and decomposition product gas analysis; Hutnické listy 12, 1096 (1957).
- 61. Šatava V., Stránský K.: Gradient furnace with defined atmospheres; Silikáty 3, 343 (1959).
- 62. Vaniš M., Koráb O.: Simple apparatus for quick DTA; Silikáty 4, 266 1960).
- 63. Blažek A., Halousek J.: *A registering thermobalance*; Silikáty 6, 100 (1962).
- 64. Jesenák V., Tölgyessy J.: *Radioactive kryptonates*; Chem. Listy *60*, 577 (1966).
- Komrska J.: Permeability thermal analysis; Silikáty 11, 51 (1967).

- 66. Chromý S.: Photoelectric apparatus for refractive index determination by the immersion method; Amer. Mineral. 54, 549 (1969).
- 67. Balek V.: Use of methods of emanation thermal analysis in the study of solid-state processes; Silikáty 13, 39 (1969).
- Proks I., Zlatkovský J.: Laboratory Techniques and Methods. Periodic Thermal Analysis; Chem. Zvesti 23, 620 (1969).
- Kolomazník K., Zapletal J., Soukup J.: Application of electronic microbalance for gravimetric thermal analysis; Chemické listy 11, 1203 (1971).
- Vepřek O., Rykl D., Šatava V.: The study of hydrothermal processes by the DTA method; Thermochim Acta 12, 7 (1974).
- 71. Šatava V.: New method of differential hydrothermal analysis DHTA; J. Amer. Cer. Soc. 58, 357 (1975).
- 72. Šatava V., Vepřek O.: Effect of the sample thermal conductivity on the calibration constant in DTA; Thermochim Acta 17, 252 (1976).
- Blažek A, Ederová J, Endrýs J.: Thermal conductivity of glass and the methods of its measurements; Silikáty 25, 359 (1981).
- Šatava V., Vepřek O.: Differential hydrothermal analysis; Stavivo 12, 68 (1981).
- Balek V.: Use of emanation TA in the evaluation of sinterability; Silikáty 27, 257 (1983).
- Balek V.: Emanation TA and its application; Silikáty 28, 147 (1984).
- 77. Hrabě Z., Svetík S.: *The application of heat flow sensor to study the hydration of inorganic binders;* Thermochim. Acta 93, 299 (1985).
- Chromý S., Hložek M.: Method of quick thermal analysis;
 Thermochim. Acta 92, 433 (1985).
- 79. Šolc Z., Trojan M.: Application possibilities for thermoelectrometry; Silikáty 29, 351 (1985).
- 80. Šolc Z., Trojan M., Kuchler M.: Thermoelectrometry as a method for reactivity estimation of solid powdery materials; Thermochim. Acta 92, 425 (1985).
- 81. Lukáč, P., Tölgyessy, J., Vaniš, M., Lapčík: *Thermal de-kryptonation characterization of some solid materials*; Thermochim. Acta 92, 429 (1985).
- 82. Lach, V.: Some applications of the decrepitating technique and thermosonimetry in research of materials. Thermochim. Acta 110, 265 (1987).
- 83. Illeková E., Ambrovič P., Czomorová K.: *Investigation* of structural relaxation of amorphous metallic alloy by thermomagnetometry. J. Thermal Anal. 32, 9 (1987).

Apparatuses of TA

- 84. Sokol L.: Automatic apparatus for DTA; Silikáty 1, 177 (1957).
- 85. Šatava V., Trousil Z.: Simple construction of apparatuses for automatic DTA; Silikáty 4, 272 (1960).
- 86. Proks I., Šiške V.: Low temperature DTA apparatus; Chem Zvesti 15, 309 (1961).
- Blažek A., Halousek J.: Instrumentation for thermogravimetry in vacuum; Silikáty 6, 100 (1962).
- Hrubý A., Beránková J.: Apparatus with a large temperature gradient for preparation of single crystals; Czech. J. Phys. A15, 740 (1965).
- 89. Němec L.: A device for measurement of electrode impedance; J. Electroanal. Chem 18, 467 (1968).
- 90. Malinger M., Brandštetr J.: Use of Czechoslovak ther-

- mistors in thermometric analysis; Chem. Listy 63, 931 (1969).
- 91. Šesták J, Burda E, Holba P, Bergstein A.: *An apparatus for DTA in vacuum and regulated atmospheres*; Chem. Listy *63*, 785 (1969).
- 92. Mráček J.: *Utilization of DTA principle for primary calibration of thermocouples*; Silikáty 15, 381 (1971).
- 93. Brandštetr J., Malinger M.: Device for differential thermometric measurements at constant temperature; Chem. Listy 66, 88 (1972).
- 94. Brown A, Šesták J, Kronberg A.: Vertical tungsten furnace for thermal studies up to 2700 °C; Czech J Phys. B23, 612 (1973).
- 95. Nývlt J., Crha J., Sura J.: A laboratory programmed thermoregulator; Chem. Listy 68, 164 (1974).
- Velišek J.: Apparatus for quantitative thermal analysis;
 Chem. Listy 68, 1185 (1974).
- 97. Rosický J., Kmoničková S.: *Apparatus for quantitative DTA*; Silikáty 20, 373 (1976).
- 98. Procházka S., Sura J., Nývlt J.: *Programmed temperature* controller used in investigation of crystallization; Chem. Listy 71, 1086 (1977).
- 99. Jesenák V., Tomková V.: Apparatus for thermal dekryptonation thermal analysis; Radiochem. and Anal. Let. 30, 89 (1977).
- Brandštetr J.: New instrument and method for enthalpiometric analysis and calibration; J. Thermal Anal. 14, 157 (1978).
- 101. Nerad I., Proks I., Zlatkovský I.: Apparatus for determining the time of passage of a solid particle through a steady temperature fields; Silikáty 28, 261 (1984).
- 102. Šolc Z., Trojan M.: Simple program control furnace; Silikáty 31, 365 (1987).

Calorimetry

- 103. Krupka F., Horák Z.: The determination of the specific heat of a liquid in an electric calorimeter; Czech. J. Phys. 46, 619 (1956).
- 104. Proks I., Eliášová M., Pach L.: Calorimeter for measurements of heats of solution; Chem. Zvesti 21, 908 (1967).
- 105. Abbrent M., Sojka B., Pekárek V.: *Isothermal differential calorimeter*. Chem. Listy *63*, 1042 (1969).
- 106. Velíšek J.: *High-temperature calorimetry*; Čs. čas. fyz. *A20*, 513 (1970).
- 107. Tydlitát V., Blažek A., Halousek J.: A copper drop-calorimeter with adiabatic shield for enthalpy measurement up to 1700 K; Czech. J. Phys. A 21, 817 (1971).
- 108. Brandštetr J., Malinger M., Kupec J.: Devise for differential thermometric (enhalpiometric) measurements; Chem. Listy 66, 88 (1972).
- Smíšek M., Rameš J., Křesťanová V.: Calorimeter for measurement of heats of evaporation on a microscale; Chem. Listy 68, 738 (1974).
- 110. Smíšek M., Křesťanová V.: *Reaction calorimeter*; Chem. Listy 68, 738 (1974).
- 111. Pekárek V.: Possibilities and present state of calorimetric experiments; Chem. Listy 69, 785 (1975).
- 112. Blaho D., Abbrehnt M., Pekárek V.: *Differential quasiisothermal calorimeter with digital output*; Chem. Zvesti 30, 621 (1976).
- 113. Proks I., Eliášová M., Zlatkovský I.: High-temperature drop calorimetry in phase analysis; Silikáty 21, 253 (1977).
- 114. Zlatkovský I.: Evaluation of calorimetric measurements

- on basis of thermal losses; Silikáty 21, 71 (1977).
- 115. Brandštetr J.: Precision of proposed double injection methods for direct enthalpiometric analysis; Coll. Czech. Chem. Comm. 42, 56 (1977).
- Brandštetr J.: New instruments and methods for enthalpiometric analysis and calibration; J. Thermal Anal. 14, 157 (1978).
- Velíšek J.: Calorimetric methods; Chemické listy 72, 801 (1978).
- 118. Proks I., Kosa I.: Evaluation of the use of dissolution calorimetry in phase analysis; Silikáty 24, 271 (1980).
- 119. Velíšek J.: High-temperature twin calorimeter for the measurement of mixing heats of alloys in solid state; Chem. Listy 75, 201 (1981).
- 120. Brandštetr J.: Present state and future development of instruments for thermometric analysis; J. Thermal Anal. 21, 357 (1981).
- 121. Hakl J.: Over-adiabatic calorimetry (OAC); Thermochim. Acta 81, 319 (1984).
- 122. Velich V., Dittrich F., Timar J.: *Isoperibolic calorimeter* with an online computer; Chem. Listy 79, 661 (1985).
- 123. Kubičár L., Illeková E.: Use of pulse method for study of structural changes of materials; Thermochim. Acta 92, 441 (1985).
- 124. Kubičár L.: Trends in methods of measurements of thermophysical properties of solids; Thermochim. Acta 110, 205 (1987)
- 125. Brandštetr J.: Outline of some new calorimetric techniques and instrumentation; Thermochim. Acta 110, 165, (1987).
- 126. Nerad I., Proks I.: Determination of equilibrium quantities of the system formed by thermal decomposition: Experimental equipment; Chem. Zvesti 41, 3 (1987).
- 127. Nerad I., Vitková S., Proks I.: New method for determination of the equilibrium state in the system involving binary reactions; J. Thermal Anal. 33, 291 (1988).

Theory of DTA/DCS

- Šatava V.: Differential thermal analysis; Silikáty 1, 207 (1957).
- 129. Proks I.: Influence of rate of temperature increase on the quantities important for the evaluation of DTA curves; Silikáty 5, 114 (1961).
- 130. Šesták J.: Temperature effects influencing kinetic data accuracy obtained by thermographic measurements under constant heating; Silikáty 7, 125 (1963).
- 131. Šesták J., Berggren G.: DTA: Use for enthalpic and kinetic measurements; Chem. listy 64, 695 (1970).
- 132. Proks I.: Effect of quantities controlling DTA on the difference between measured and theoretical temperatures; Silikáty 14, 287 (1970).
- 133. Hrubý A.: Evaluation of glass forming tendency by means of DTA; Czech J. Phys. B22, 1187 (1972) and B23, 1623 (1973).
- 134. Svoboda H., Šesták J.: *A new approach of DTA calibration by predetermined amount of Joule heat*; In "Thermal Analysis" (proceedings of 4th ICTA,, E. Buzagh, Ed.) Akademia Kiado, Budapest 1974, pp.726.
- 135. Nevřiva M., Holba P., Šesták J.: Application of DTA in determination of transformation heats; Silikáty 20, 33 (1976).
- 136. Šesták J., Holba P., Bárta R.: Theory and practice of TA methods based on the indication of enthalpy changes; Silikáty 20, 83 (1976).

- 137. Šesták J., Holba P., Lombardi G.: *Quantitative evaluation* of thermal effects: theory and practice; Annali di Chimica Roma 67, 73 (1977).
- 138. Šatava V, Vepřek O.: Effect of sample thermal conductivity on the calibration constant in DTA; Thermochim. Acta 17, 252 (1977).
- 139. Hrubý A.: Study of glass formation applicability and phase diagram; J. Non-cryst. Solids 28, 139 (1978).
- Nevřiva M.: Mn-Cr-O solid solutions as materials for thermoanalytical calibration; Thermochim. Acta 22, 187 (1978).

Thermodynamics and phase equilibria

- 141. Šatava V.: Contemporary foresight on the structure of glasses; Sklář a keramik 13, 6 (1956).
- 142. Holba P., Pollert E., Kitzinger E.: Experimental arrangement for investigation of solid-gas equilibria; Silikáty 13, 271 (1969).
- 143. Holba, P.: On calculations of activities of chemical individuals from disorder models; Thermochim Acta 3, 475 (1972).
- 144. Šatava V.: Nature of vitreous state and conditions of glass-formation; Czech J Phys A23, 565 (1973).
- 145. Kratochvíl J.: *Rational thermodynamics*; Czech J Phys *A23*, 1 (1973).
- 146. Malinovský M.: Concentration vectors in phase diagrams; Chem. Zvesti 28, 463 (1974).
- 147. Malinovský M.: *Criteria of thermodynamic consistency*; Chem. Zvesti 28, 489 (1974).
- 148. Malinovský M.: Equation of liquidus curve in eutectic systems; Chem. Zvesti 30, 721 (1976).
- 149. Hrma P.: *Modern Approach to Classical Thermodynamics*; Chem. listy *69*, 1229 (1975).
- 150. Holba P.: Determination of binary phase diagram curve using invariant point data; Silikáty 20, 193 (1976).
- 151. Holba P.: *Thermodynamic aspects of thermal analysis*; Silikáty 20, 45 (1976).
- 152. Šesták J.: Thermodynamic basis for the theoretical description and correct interpretation of thermoanalytical experiments; Thermochim. Acta 28, 197 (1979).
- 153. Holba P.: Enthalpy and phase relations at melting in heterogeneous systems; Silikáty 23, 289 (1979).
- 154. Hrma P.: *Thermodynamics of batch melting*; Glastech. Ber. 55, 138 (1982).
- 155. Šesták J.: *Thermodynamic aspects of glassy state*; Thermochim Acta 95, 459 (1985).
- Nevřiva M., Šesták J.: On the study of solid-liquid stablemetastable phase equilibria; Thermochim. Acta 92, 623 (1985).
- 157. Šatava V.: Determination of standard enthalpies, Gibbs energies and entropies of formation of hydrated calcium sulphoaluminates; Silikáty 30, 319 (1986).
- 158. Majling J., Jesenák V.: Algorithmizing the calculation of equilibria phase composition of multicomponent systems; Silikáty 30, 319 (1986)
- 159. Sopková A.: *Inorganic complexes*; Thermochim. Acta 110, 389 (1987).
- 160. Šesták J., Chvoj Z.: Thermodynamics and thermochemistry of kinetic (real) phase diagrams involving solids; J. Thermal Anal. 32, 1645 (1987).
- 161. Šatava, V.: Direct determination of standard enthalpies and Gibbs energies of formation and absolute entropies of hydrated calcium sulphoaluminates and carboaluminates;

- Thermochim. Acta 132, 285 (1988).
- Liška M., Daněk V.: Computer calculation of the phase diagrams of silicate systems. Ceramics-Silikaty 34, 215 (1990).

Kinetics

- Šatava V., Körbel J.: Kinetics of thermal decomposition of silver permanganate; Collect. Czech. Chem. Com. 22, 1380 (1957).
- 164. Šatava V.: Reactions of solids; Silikáty 4, 67 (1960).
- Šatava V.: Reaction of solids with liquids; Silikáty 5, 171 (1961).
- 166. Šatava V., Marek J., Matoušek J.: Dehydration kinetics of α- a β- calcium sulphate hemihydrates in suspension; Silikáty 5, 309 (1961).
- 167. Šatava V., Šesták J.: Kinetic analysis of thermogravimetric data; Silikáty 8, 134 (1964).
- 168. Šesták J.: Errors of kinetic data obtained from TG curves at increasing temperature; Talanta 13, 567 (1966).
- 169. Hulínský V., Šatava V.: *Gypsum solubility within temperatures* 100-140 °C; Silikáty 11, 47 (1967).
- Šesták J.: Review of kinetic data evaluation from nonisothermal and isothermal TG data; Silikáty 11, 153 (1967).
- 171. Šesták J., Šatava V., Řihák V.: Algorithm for kinetic data computation from thermogravimetric data obtained at increasing temperature; Silikáty 11, 153 (1967).
- 172. Procházka S.: Surface area changes during sintering; Amer. Cer. Soc. Bull. 47, 753 (1968).
- 173. Vašková L, Hlaváč J. Crystallization of quartz glass; Silikáty 13, 211 (1969).
- 174. Šatava V., Škvára F.: Mechanism and kinetics of solidstate reactions; J. Am. Ceram. Soc. 52, 591 (1969).
- 175. Šatava V.: Mechanism and kinetics of crystallization from nonisothermal measurements; Thermochim. Acta 2, 423 (1971)
- 176. Vachuška J., Vobořil M.: Kinetic data computation from non-isothermal thermogravimetric curves of non-uniform heating rate; Thermochim. Acta 2, 379 (1971).
- 177. Šesták J., Berggren G.: Study of the kinetics of the mechanism of solid-state reactions at increasing temperatures; Thermochim. Acta 3, 1 (1971).
- 178. Šimon P.: Fourty years of the Sestak-Berggren equation; Thermochim. Acta 520, 156 (2011).
- 179. Holba P., Šesták J.: *Kinetics with regard to the equilibrium of processes studied by non-isothermal techniques*; Zeit. physik. Chem. N.F. 80, 1 (1972).
- Mianowski A.: Consenquences of Holba-Sestak equation;
 J. Thermal Anal. Calor. 96, 507 (2009).
- 181. Šesták J., Kratochvil J.: *The role of state constitutive equations in chemical kinetics;* Thermochim. Acta 7, 330 (1973).
- 182. Šatava V., Šesták J.: Kinetics and mechanism of thermal decomposition at iso- and non-iso- thermogravimetry; Anal. Chem. 45, 153 (1973).
- 183. Šatava V.: Fundamental principles of kinetic data evaluation from TA curves; J. Thermal Anal. 5, 217 (1973).
- 184. Kratochvíl J., Šesták J.: Rational approach to thermodynamic processes and constitutive equations in iso- and non-isothermal kinetics; J. Thermal Anal. 5, 193 (1973).
- 185. Šesták J.: Applicability of DTA to the study of crystallization kinetics; Phys. Chem. Glasses 15, 137 (1974).
- 186. Matuchová M., Nývlt J.: Theory of crystallization rate;

- Chem. Listy 69, 1 (1975).
- 187. Holba P., Nevřiva M., Šesták J.: Analysis of DTA curve and related calculation of kinetic data using computer technique; Thermochim. Acta 23, 223 (1978).
- 188. Kubíček P., Leško J.: Determination of the kinetic parameters from non-isothermal measurements with a general temperature program; Thermochim. Acta 31, 21 (1979).
- 189. Šesták J.: *Philosophy of nonisothermal kinetics*; J. Thermal Anal. *16*, 503 (1979).
- 190. Šimon P., Valko L.: Autocatalytic effect of hydrogen chloride on the thermal dehydrochlorination of polyvinyl chloride; Chemické Zvesti 37, 581 (1983).
- 191. Illeková E.: On the Various Activation Energies at Crystallization of Amorphous Metallic Materials; J. Non-Cryst. Solids 68, 153 (1984).
- 192. Nývlt J.: *Crystal growth measurements*; Cryst. Res. Tech. *18*, 1461 (1983).
- 193. Jesenák V.: Philosophy of the mechanism of diffusion controlled processes; Thermochim. Acta 92, 39 (1985).
- 194. Jesenák V.: Thermal effects of oscillating solid-state reactions; Thermochim. Acta 85, 91 (1985).
- 195. Kemény T., Šesták J.: Comparison of crystallization kinetics determined by isothermal and non-isothermal methods; Thermochim. Acta 110, 113 (1987).
- 196. Málek J.: The crystallization of $Ge_{40}S_{60}$ glass. Thermochim. Acta 129, 293 (1988).
- 197. Militký J., Málek J., Šesták J.: Parameter distortion by unappropriate nonisothermal treatment; J. Thermal Anal. 35, 1837 (1989).
- 198. Chvoj Z., Kožíšek Z., Šesták J.: Nonequilibrium processes of melt solidification during programmed temperature changes and the metastable phases formation; Thermochim. Acta 153, 349 (1989).
- 199. Šimon P.: Kinetics of polymer degradation involving the splitting-off of small molecules; Polym. Degrad. Sta. 29, 155 (1990).

Mechanical and transport properties

- 200. Bárta R., Šatava V, Procházka S, Hlaváč J: *Základní výzkum silikátů;* (Basic Research of Silicates), SNTL Praha 1957.
- 201. Šatava V., Šesták J.: Influence of setting temperature on the structure and strength of the plaster of Paris; Silikáty 6, 178 (1962).
- 202. Hrubý A., Kubelík I.: *Electrical conductivity and thermoelectric power in semiconductors*; Czech. J. Phys. B *15*, 740 (1965).
- 203. Hrma P.: Corrosion of refractory materials by molten glass; Silikáty 13, 165 (1969).
- 204. Hlaváč J., Matoušek J.: Diffusion in molten oxide silica glasses; Silikáty 15, 333 (1971).
- 205. Matoušek J., Hlaváč J.: Study of volatilization of lead glasses; Glass Technol. 12, 103 (1971).
- 206. Kaščejev I., Matěj J., Bartuška M.: Corrosion of mullite ceramics by binary glass under free convection; Silikáty 16, 25 (1972).
- 207. Němec L.: Dissolution and diffusion of technologically significant gases in glass; Silikáty 16, 347 (1972).
- 208. Hrma P., Šatava V.: *Model for strength of brittle porous materials;* J. Am. Ceram. Soc. *57*, 71 (1974).

- 209. Hrma P.: Diffusion theory of ceramic body growth; Silikáty 22, 357 (1978).
- 210. Kubíček P., Wozniaková B., Leško J.: *Determination of diffusion coefficients with a general temperature program using DTA;* Thermochim. Acta *64*, 229 (1983).
- 211. Illeková E.: Model of mass transport in amorphous metallic materials; Acta Phys. Slov. 34, 255 (1984).
- 212. Šajgalík P, Haviár M, Pánek Z, Contribution to the sintering diagrams; Zeits. Metall. 77, 193 (1986).
- 213. Liška M., Hamlík L., Kanclíř E.: *Enamel viscosity dependence on modeling compositional and thermal relations*; Silikáty *31*, 43 (1987).
- 214. Málek J., Klikorka J., Tichý L.: Conductivity measurements during crystallization of glasses; Mater. Sci. Let. 5, 183 (1986).
- 215. Mareš J., Krištofik J., Šmíd V.: On the conductivity in semiinsulating semiconductors; Sol. State Com. 60, 275 (1986).
- 216. Málek J., Klikorka J., Šesták J., Tříska A.: *Thermoelectrical conductivity as a complementary method to DTA;* Thermochim. Acta *110*, 281 (1987).
- 217. Helebrant A., Matoušek J.: Mathematical models of the interaction of glass with water and with aqueous solutions; Silikáty 32, 173 (1986).
- 218. Šatava V.: Strength and microstructure of cast gypsum; Ceramics-Silikáty 40, 72 (1996).

History and nomenclature

- 219. Holba, P., Šesták J.: On the nomenclature of thermoanalytical methods associated with energy changes; Thermochim. Acta 13, 471 (1975).
- 220. Šesták J., Holba P., Fajnor V.: System of Czech and Slovak terms used in thermal analysis; Chem. Listy 77, 1292 (1983).
- 221. Hlaváč J.: The present state and perspective view of the material research; Silikáty 29, 169 (1985).
- 222. Mackenzie R., Proks I.: Comenius and Black as progenitors of thermal analysis; Thermochim. Acta 92, 3 (1985).
- 223. Šesták J., Mackenzie R.C.: *Bárta Rudolf (1897-1985)*; J. Thermal Anal. *31*, 3 (1986).
- 224. Nevřiva M, Rosický J, Proks I, Kanclíř E.: *Pioneers of thermal analysis in Czechoslovakia*; Thermochim. Acta *110*, 553 (1987).
- 225. Bartuška M., Götz J.: *The Silikáty periodical in future*; Silikáty *33*, 289 (1989).
- 226. Proks I.: Evaluation of the Knowledge of Phase Equilibria; In book "Kinetic Phase Diagrams: nonequilibrium phase transitions" (Z. Chvoj, J. Šesták, A. Tříska, eds.), p.1-49, Elsevier, Amsterdam 1991.
- 227. Proks I. *Celok je jednoduchší ako jeho časti;* (Whole is simpler than its parts), Publ. House of Slovak Academy of Sciences, Bratislava 2012.